

Stationary Phase Selectivity

When Purchasing a new column, the most important consideration is the stationary phase. There are many different interactions that occur between the analytes and the functional groups of the stationary phase. These interactions contribute more to the overall results of the analysis than any other factor in the column. That is why it is important to understand as much about your column and sample as possible.

Table I shows the chemical structure of most common stationary phases. Changes in selectivity can be observed by using a column with different functional groups as well as increasing the percentage of substitution of those functional groups. The non-polar Rtx'-1 phase will preferentially retain non-polar compounds compared to polar compounds such as alcohols. As non-polar methyl units are substituted with polar functionalities such as phenyl and cyanopropyl units, the selectivity of the column shifts towards more polar compounds. In turn, non-polar compounds are retained less as there are less overall methyl units for the nonpolar compounds to interact with. The Rtx'-200 stationary phase contains trifluoropropyl units which provide high selectivity for analytes containing lone pair electrons, such as nitro and carbonyl groups. Polyethylene glycol columns, such as Stabilwax and Rtx''/MXT-WAX columns, are polar and are highly selective towards polar compounds such as alcohols.

Table I

Comparison of structures, polarities, properties, and uses for each capillary column phase listed in order of increasing polarity.

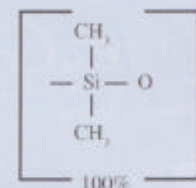

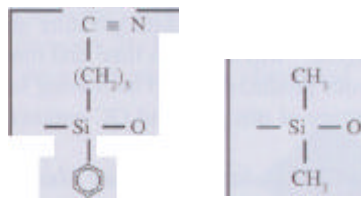


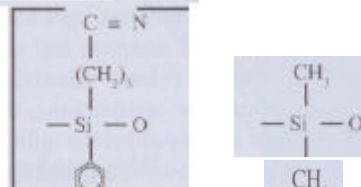
<p>Rtx[®]/MXT[®]-1 100% dimethyl polysiloxane</p>  <p>100%</p> <p>Polarity: non-polar Uses: solvents, petroleum products, pharmaceutical samples, waxes</p>	<p>Rtx[®]/MXT[®]/XT1[®]-5 5% diphenyl - 95% dimethyl polysiloxane</p>  <p>5% 95%</p> <p>Polarity: non-polar Uses: flavors, environmental samples, aromatic hydrocarbons</p>	<p>Rtx[®]/MXT[®]-1301, Rtx[®]/MXT[®]-624 6% cyanopropylphenyl 94% dimethyl polysiloxane</p>  <p>6% 94%</p> <p>Polarity: slightly polar Uses: volatile compounds, insecticide solvents in pharmaceutical prep</p>
<p>Rtx[®]/MXT[®]-20 20% diphenyl - 80% dimethyl polysiloxane</p>  <p>20% 80%</p> <p>Polarity: slightly polar Uses: volatile compounds, alcohols</p>	<p>Rtx[®]/MXT[®]-35 35% diphenyl - 65% dimethyl polysiloxane</p>  <p>35% 65%</p> <p>Polarity: intermediately polar Uses: pesticides, Aroclor[®] samples, amines, nitrogen containing herbicides</p>	<p>Rtx[®]/MXT[®]-1701 14% cyanopropylphenyl 86% dimethyl polysiloxane</p>  <p>14% 86%</p> <p>Polarity: intermediately polar Uses: pesticides, Aroclor[®] samples, alcohols, oxygenates</p>

Table I, listing column phase structures, is continued on page 7. (cont.)

Table I, listing column phase structures, continued from page 6.

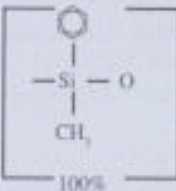
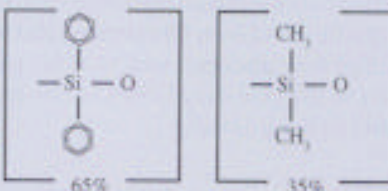
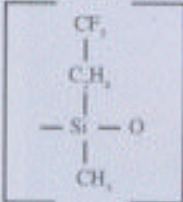
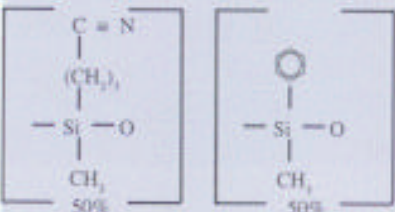
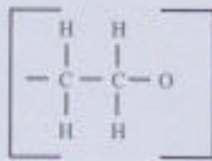
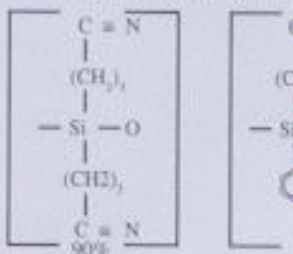
<p>Rtx[®]/MXT[®]-50 50% phenyl - 50% methyl polysiloxane</p>  <p>100%</p> <p>Polarity: <i>intermediately polar</i> Uses: <i>triglycerides, phthalate esters, steroids, phenols</i></p>	<p>Rtx[®]/MXT[®]-65TG 65% diphenyl - 35% dimethyl polysiloxane</p>  <p>65% 35%</p> <p>Polarity: <i>intermediately polar</i> Uses: <i>triglycerides, acetic acids, free fatty acids</i></p>	<p>Rtx[®]/MXT[®]-200 trifluoropropylmethyl polysiloxane</p>  <p>Polarity: <i>selective for lone pair electron</i> Uses: <i>environmental samples, active samples, drugs, ketones, alcohols</i></p>
<p>Rtx[®]-225 50% cyanopropylphenyl 50% phenylmethyl polysiloxane</p>  <p>50% 50%</p> <p>Polarity: <i>polar</i> Uses: <i>FAMEs, carbohydrates</i></p>	<p>Stabilwax[®]/MXT[®]-WAX Carbowax[®] PEG</p>  <p>Polarity: <i>polar</i> Uses: <i>FAMEs, flavors, acids, amines, solvents, xylene isomers</i></p>	<p>Rtx[®]-2330 90% biscyanopropyl 10% cyanopropylphenyl polysiloxane</p>  <p>90% 10%</p> <p>Polarity: <i>very polar</i> Uses: <i>FAMEs, ketones & diols, mono acids</i></p>

Table 11 shows retention indices for the stationary phases shown in Table I. Retention indices are mathematical derivations indicating the elution point of a probe with respect to two hydrocarbons. For example, if the retention index for benzene was 650, then it would elute halfway between C6 (RI=600) and C7 (RI=700),

EI Table II

The retention indices for each phase illustrate the differences in selectivity for a variety of compounds.

Phase	Benzene	Butanol	Pentanone	Nitropropane
Rtx [®] /MXT [®] -1	651	651	667	705
Rtx [®] /MXT [®] -5/ XTI [®] -5/Rtx [®] -5MS	667	667	689	743
Rtx [®] /MXT [®] -1301/624	689	729	739	816
Rtx [®] /MXT [®] -20	711	704	740	820
Rtx [®] /MXT [®] -35	746	733	773	867
Rtx [®] /MXT [®] -1701	721	778	784	881
Rtx [®] /MXT [®] -50	778	769	813	921
Rtx [®] /MXT [®] -65TG	794	779	825	938
Rtx [®] /MXT [®] -200	738	758	884	980
Rtx [®] -225	847	937	958	958
Stabilwax [®] /MXT [®] -WAX	963	1158	998	1230